

# From CO<sub>2</sub> to Sustainable Energy Carriers - Challenges and Progress

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*Chair of Physical Chemistry*

## ❖ Introduction

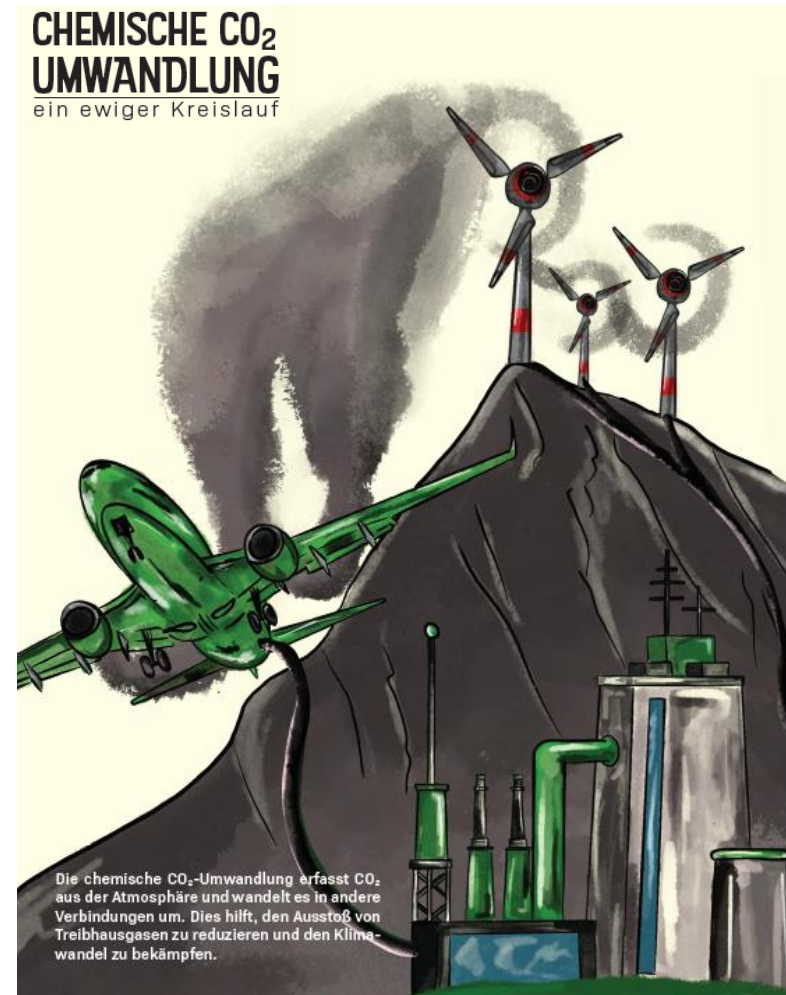
- Thermodynamic Background
- CO<sub>2</sub> Activation
- Energy Requirements

## ❖ Example for CO<sub>2</sub> Utilization

- Perovskite as Ideal Catalysts
- Reverse Water Gas Shift

## ❖ Future Development

- Catalyst Scale-Up



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# Challenge of CO<sub>2</sub> Activation

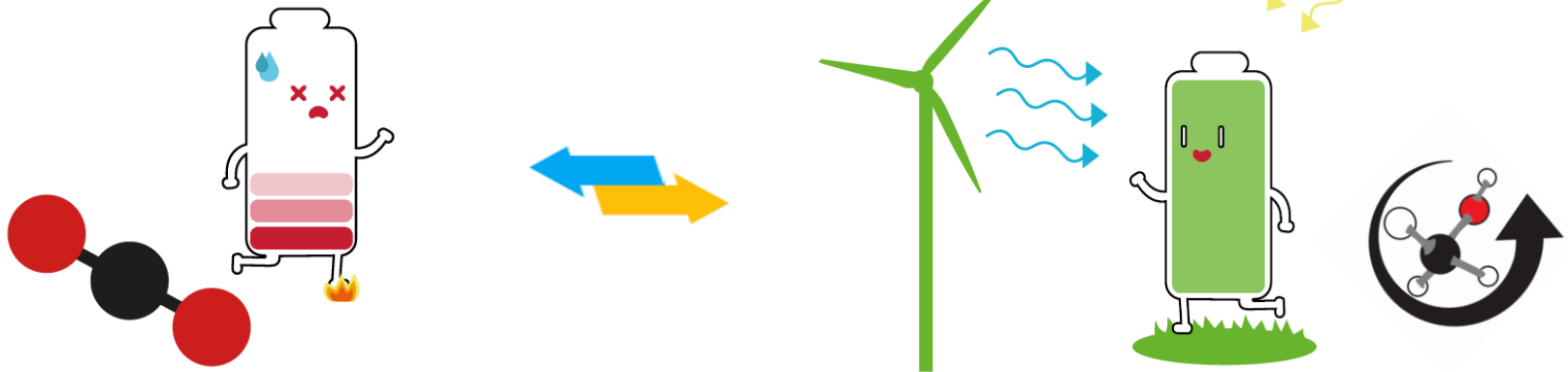


Thermodynamics:

CO<sub>2</sub> Enthalpy of formation: ~ -393,5 kJ/mol

H<sub>2</sub>O Enthalpy of formation: ~ -285,8 kJ/mol

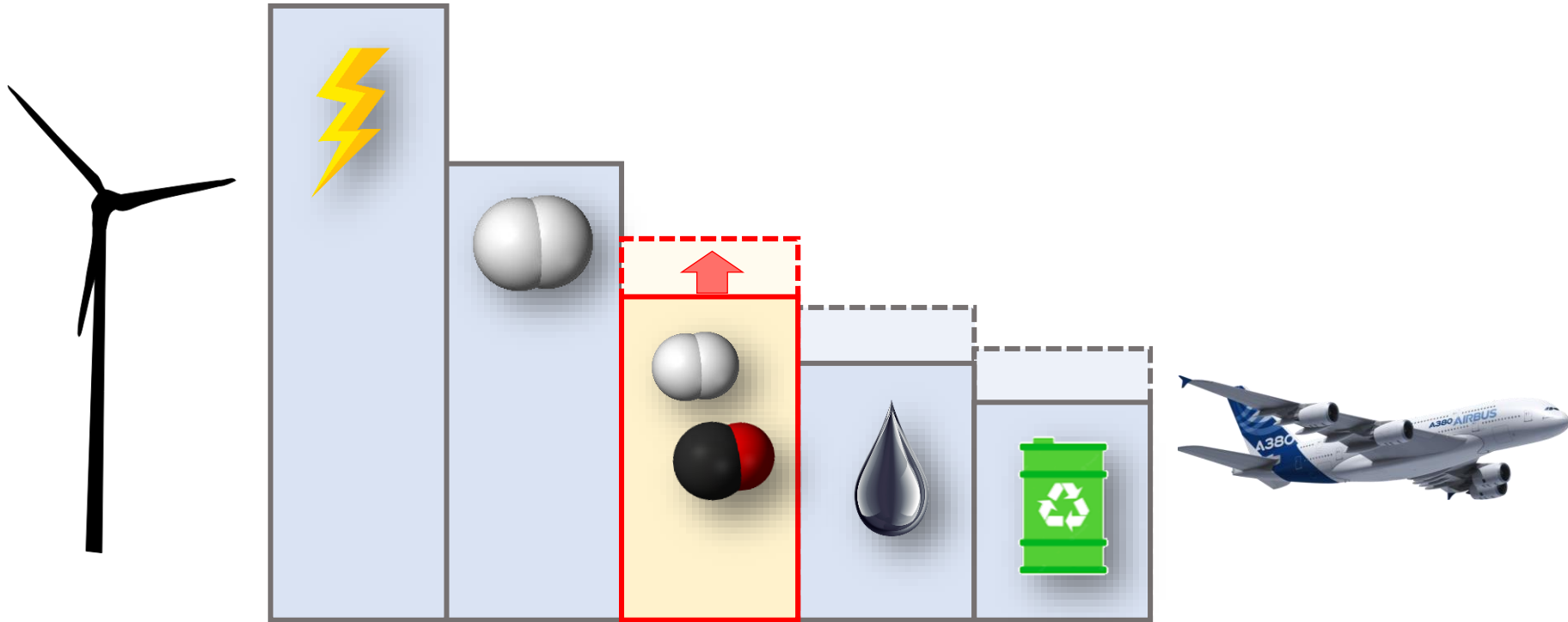
Enthalpy of combustion of e.g. methane: ~ -890.3 kJ/mol



CO<sub>2</sub> is a chemical waste  
with almost no energy

CO<sub>2</sub> conversion with renewable  
energy and hydrogen

# Chemical Energy Conversion



Valorisation of CO<sub>2</sub> is an energy intensive tasks. It is crucial to maximize the efficiency of every process step:

→ Hydrogen → CO<sub>2</sub> Conversion → FT Synthesis → Refinery

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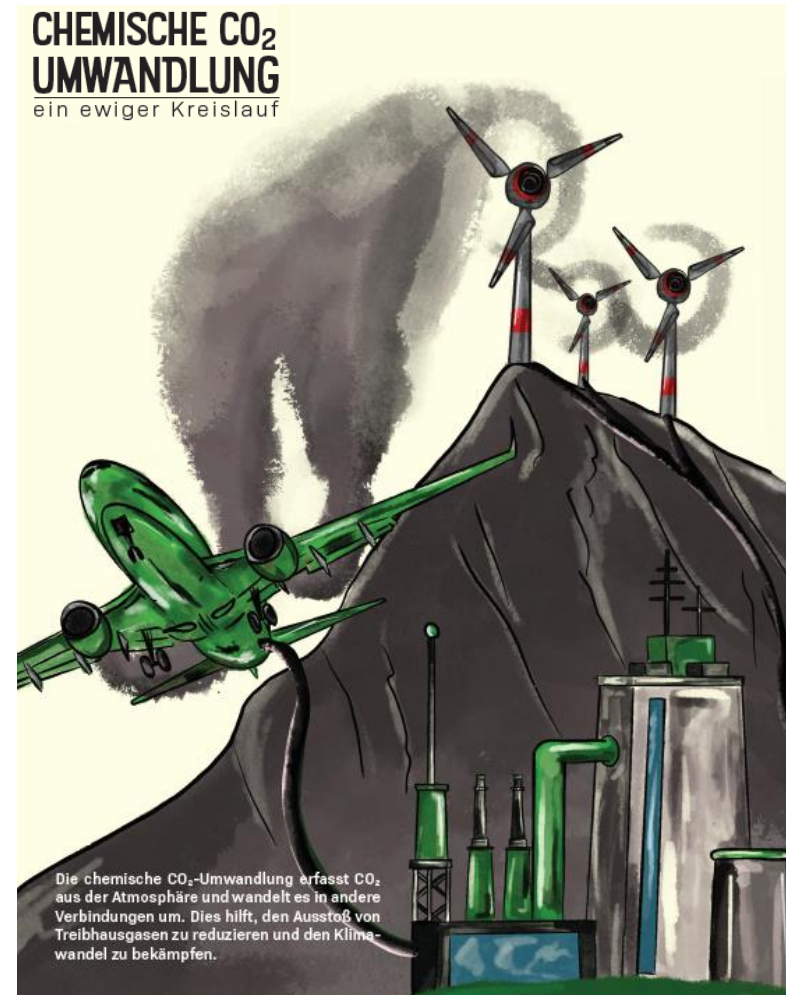
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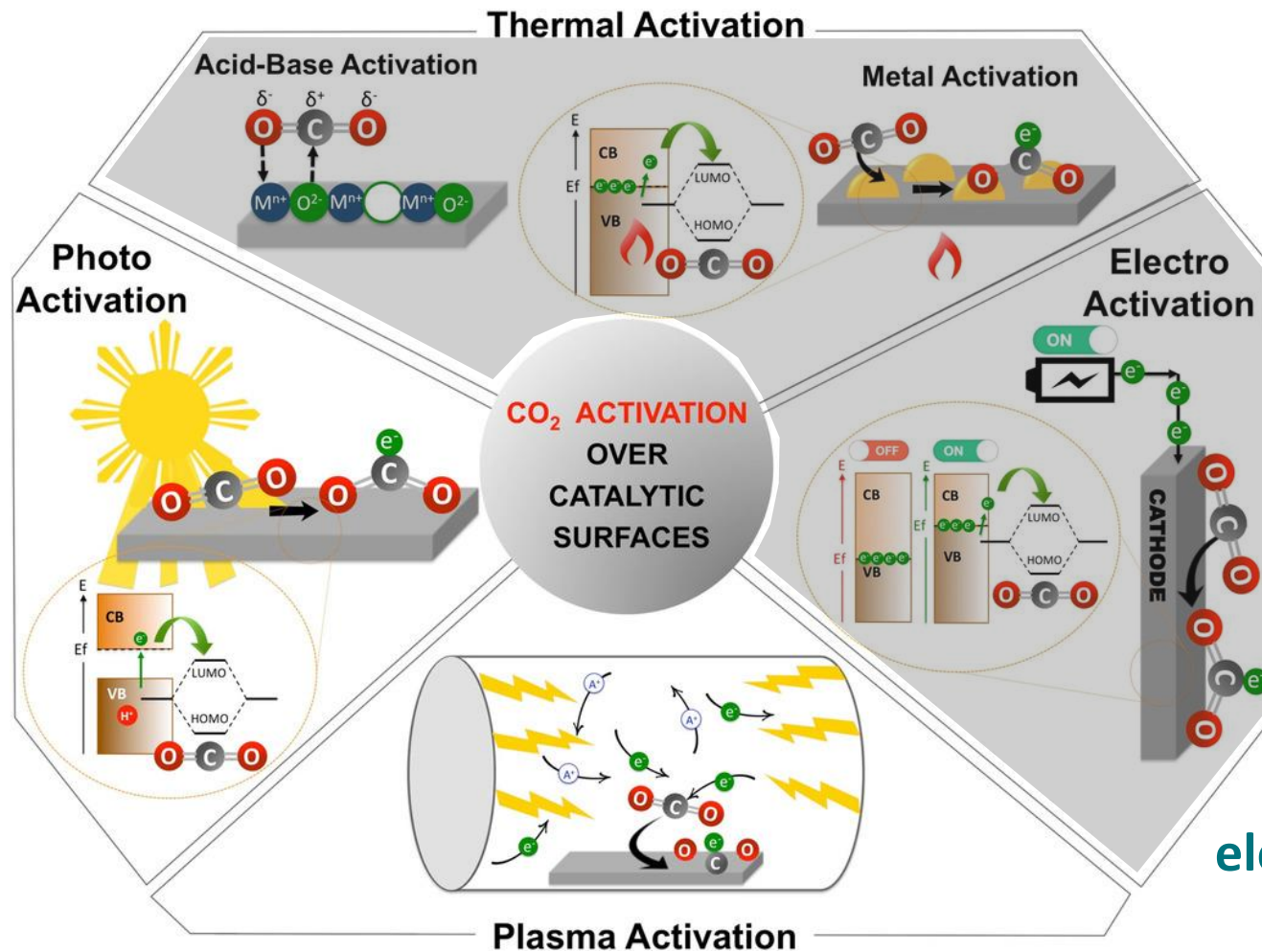
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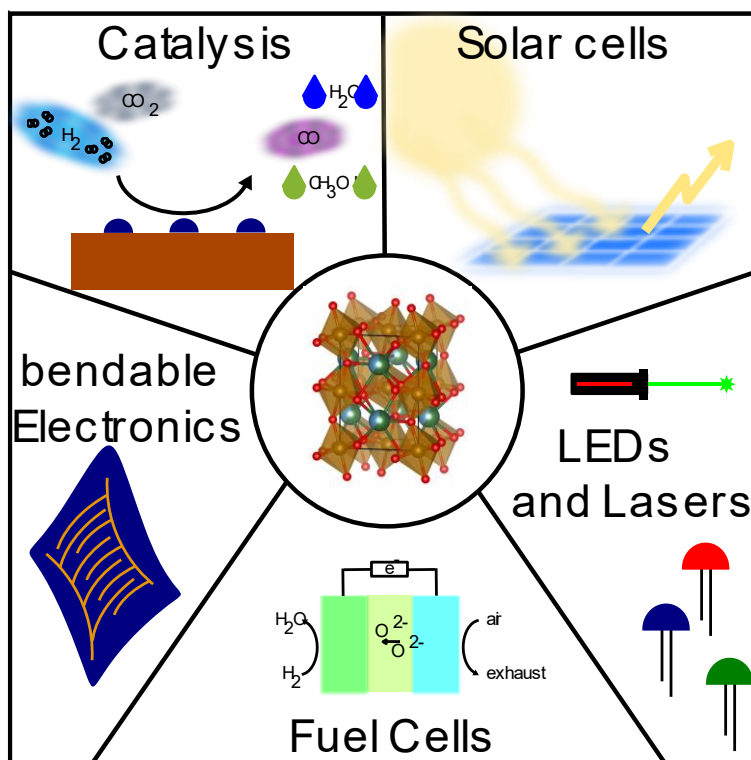
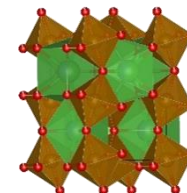
# CO<sub>2</sub> Activation

Activation required: bending, inducing partial charges or electron transfer



Combining catalysis and electrochemistry

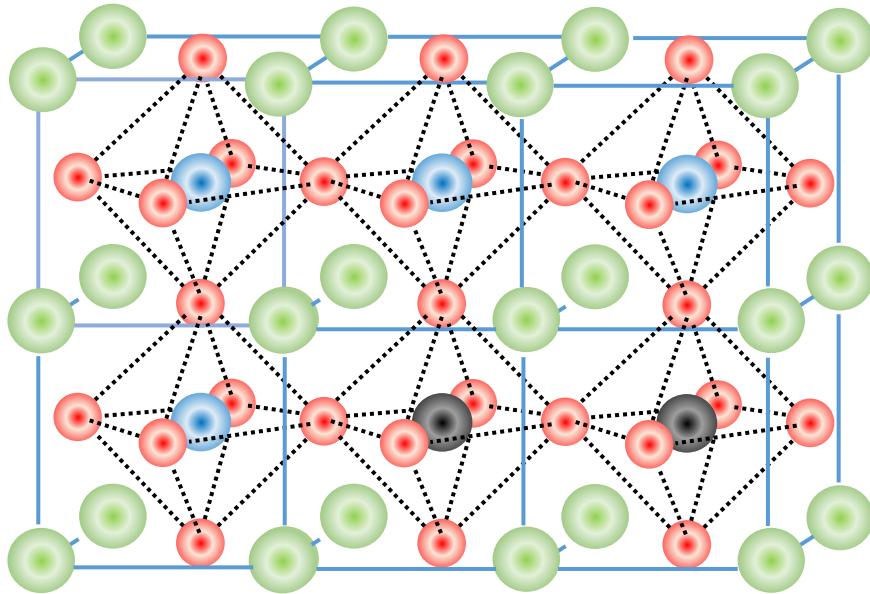
- **Material** → **Perovskite Oxides**



**Applications:**

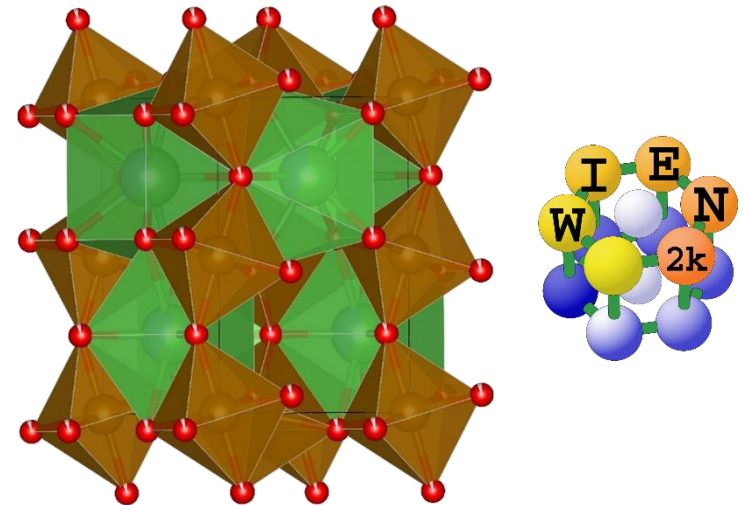
**Solid Oxide Electrolyser Cell**  
**Solid Oxide Fuel Cell**  
**Heterogeneous Catalysis**  
**Electrocatalysis**

# Doped Perovskites: $ABO_3$



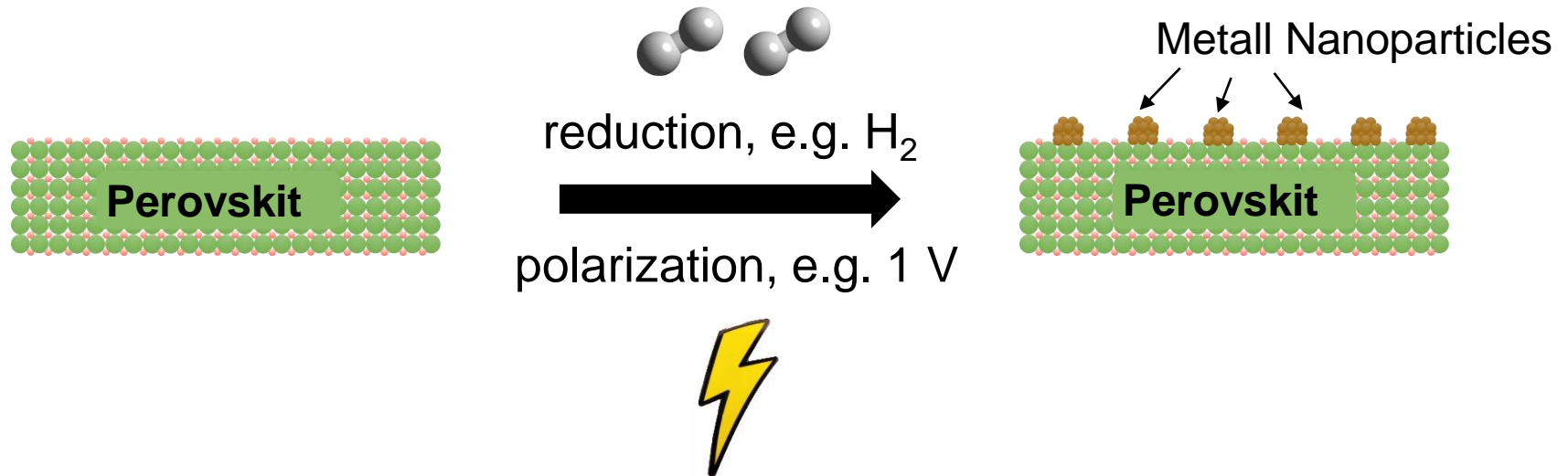
**B Site can be doped  
with catalytically  
active elements**

**Tailored for high temperature  
 $CO_2$  conversion:     $\rightarrow$  rWGS  
                                  $\rightarrow$  DRM**



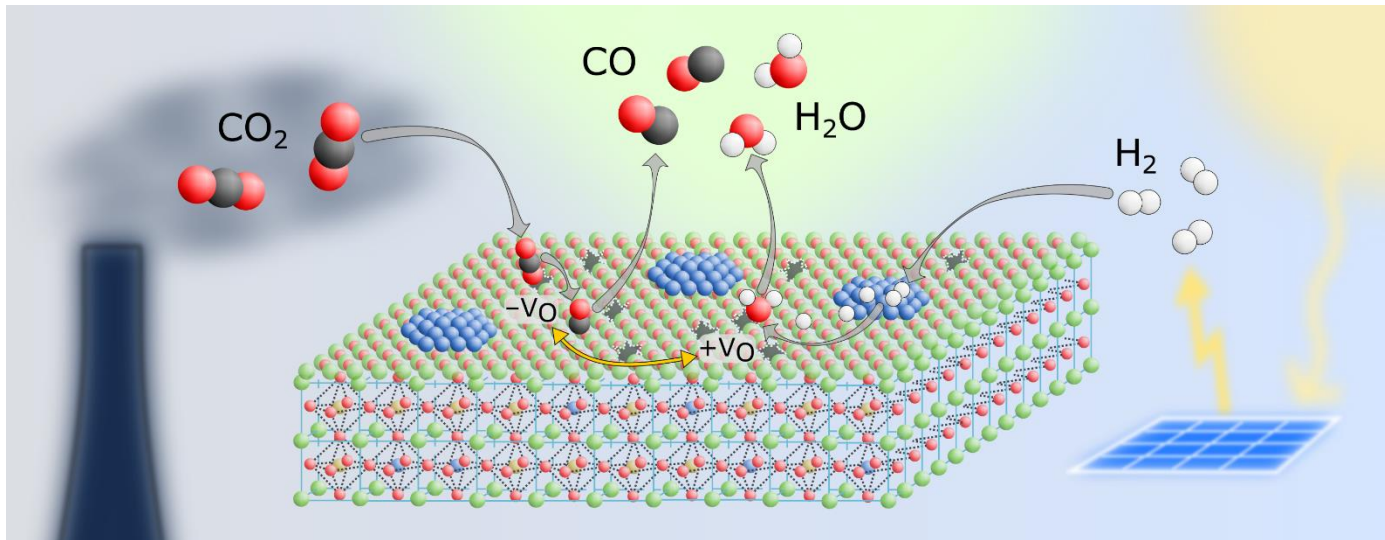
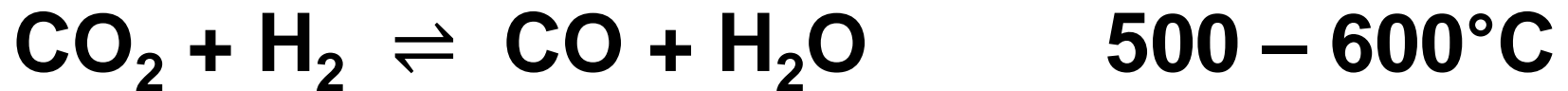


## Exsolution:



Migration of dopants to the surface → Nanoparticles

## Reverse Water-Gas Shift Reaction

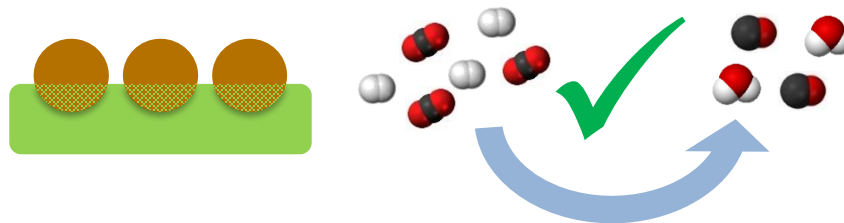
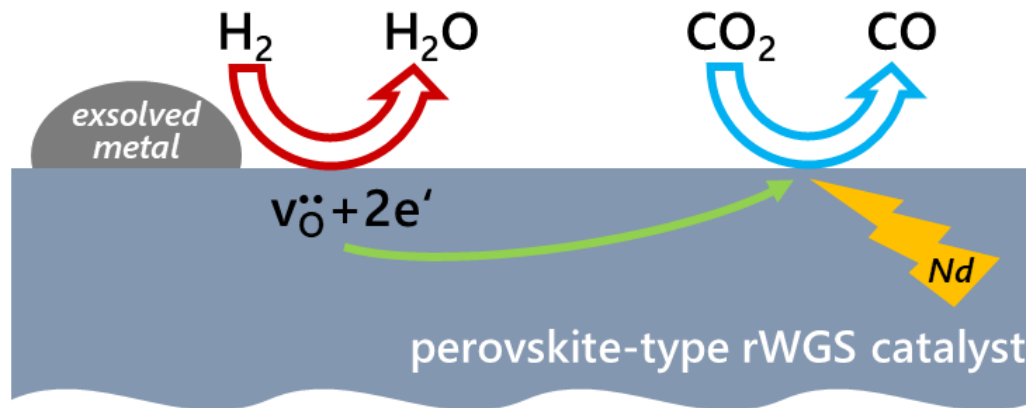


# CO<sub>2</sub> Activation

Job sharing between exsolved nanoparticles and perovskite lattice

→ H<sub>2</sub> splitting on nanoparticles, spill over to create oxygen vacancies

→ CO<sub>2</sub> activation at oxygen vacancies



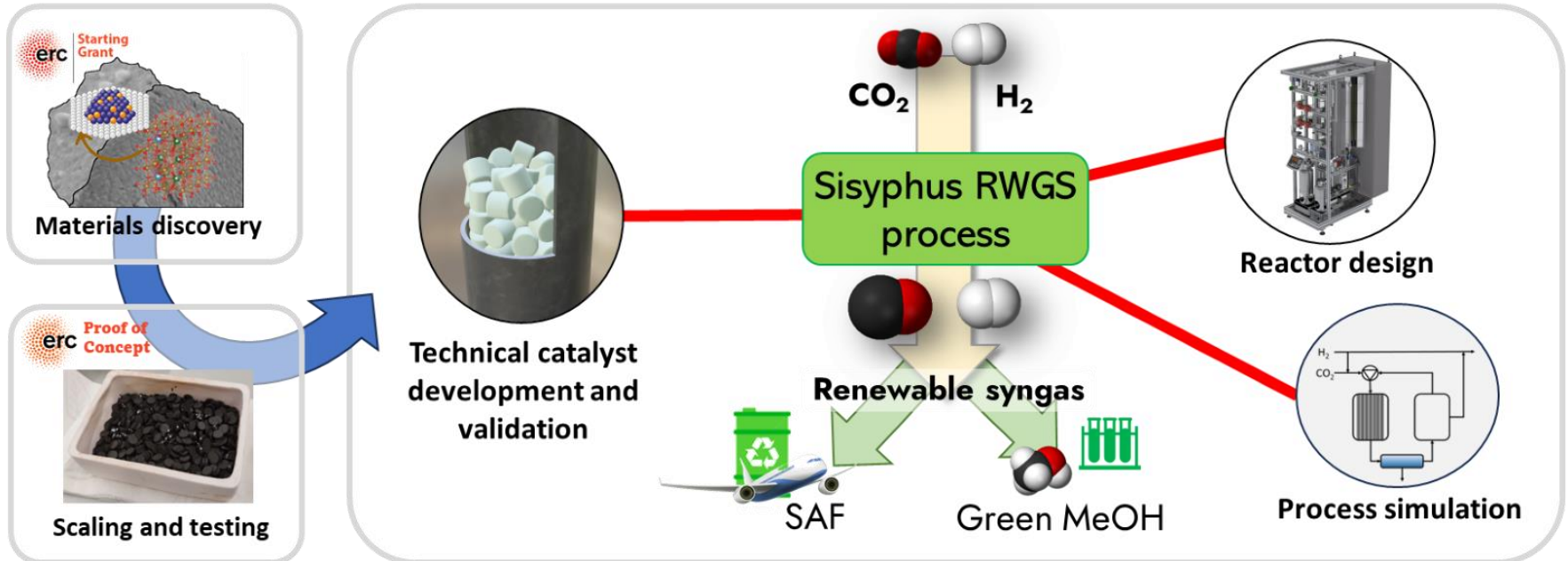
# University Spin Off



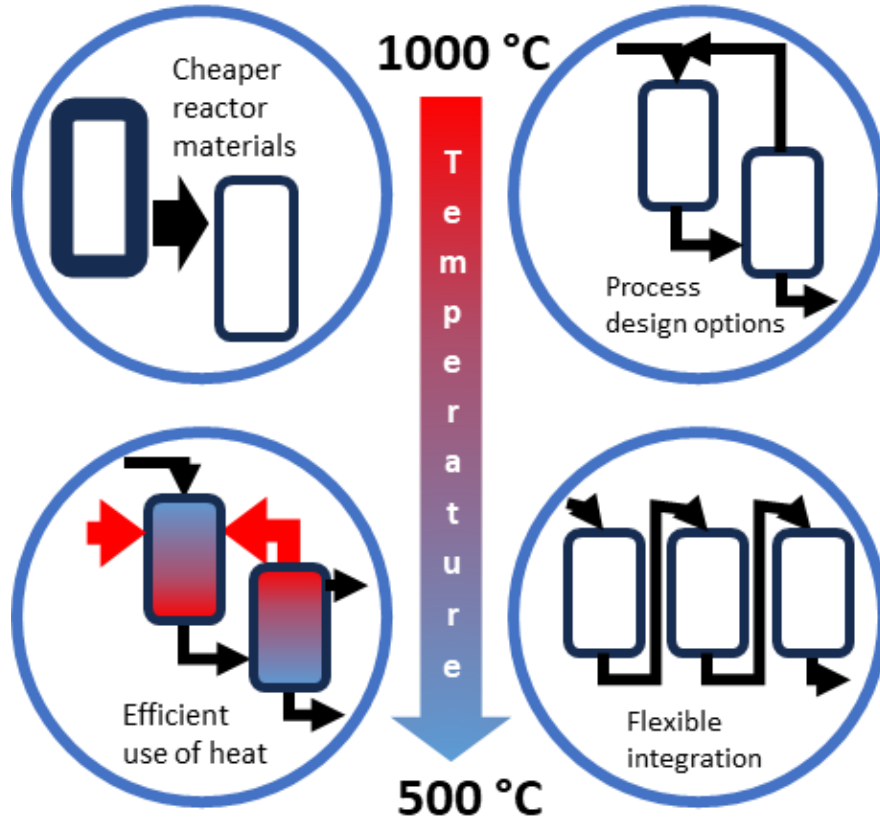
# S I S Y P H U S

Recycling Carbon Forever

Dr. Tom Cotter  
Dipl. Ing. Lorenz Lindenthal



# University Spin Off



→ Biomass upgrade to renewable syngas

→ Sustainable syngas for green methanol and e-fuels (SAF)



**FFG**

Promoting Innovation.

**FFG – Spin Off Fellowship**

Unique catalyst technology allows for lower working temperature with confirmed superiority over state-of-the-art catalysts with respect to byproduct formation and activity.

wirtschafts  
agentur  
wien

Gründungsstipendium



# Acknowledgement



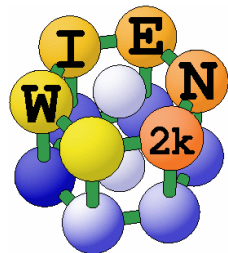
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Andreas Nenning



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Thomas Ruh  
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Stefan Löffler

Werner Artner

